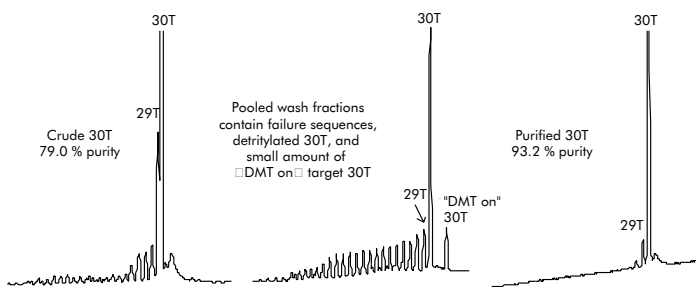


## "DMT ON" PURIFICATION OF DNA OLIGONUCLEOTIDES < 35 MER USING OASIS<sup>®</sup> HLB SPE PRODUCTS

OASIS <sup>®</sup> HLB EXTRACTION METHOD		96-well plate 30 mg WAT058951	3 cc cartridge 60 mg WAT094226	6 cc cartridge 200 mg WAT106202		
STEP	SYNTHESIS SCALE	0.1-0.2 μmol	0.2 μmol	0.1-1 μmol		
1	CONDITION: ACN	Organic solvent wets (conditions) the sorbent and frits	1 ml	2 ml	2 ml	GRAVITY FLOW
2	EQUILIBRATION: 0.1 M TEAA, pH 7	Removes ACN and equilibrates sorbent with TEAA	1 ml	2 ml	2 ml	
3	SAMPLE LOAD: in 0.1 M TEAA, pH 7	Retains target oligonucleotide and failure sequences	1 ml	2 ml	3 ml	
4	WASH 1: 8 % ACN in 0.1 M TEAA, pH 7 (v:v)	Remove weakly retained failure sequences	1 ml	2 ml	3 ml	VACUUM FLOW 1-2 ml/min
5	WASH 2: 12 % ACN in 0.1 M TEAA, pH 7 (v:v)	Remove strongly retained failure sequences	1 ml	2 ml	3 ml	
6	DETRITYLATION: 2 % TFA Apply half of the volume by vacuum, release vacuum, wait 1 min, then resume vacuum	On cartridge cleavage of DMT group from target oligonucleotide	1 ml	2 ml	3 ml	VACUUM FLOW 1-2 ml/min
7	ELUTION: 20 % ACN in 0.36 M TEAA, pH 11.3	Neutralizes TFA, dissolves and elutes target oligonucleotide	1 ml	2 ml	2 ml	

### CAPILLARY GEL ELECTROPHORESIS ANALYSIS OF FRACTIONS FROM OLIGODEOXYTHYMINE (30-mer) SPE PURIFICATION



0.1 M TEAA, pH 7 Buffer – commercially available

For 100 mL of 0.36 M TEAA buffer:

Mix 94.5 mL of MilliQ water and 0.5 mL of glacial acetic acid.

While mixing slowly add 5 mL of TEA, mix until it dissolves.

pH of final 0.36 M solution is approximately 11.3 (desirable values are between 10.8-11.5)\*

\* Keep in closed polypropylene bottle. Handle in hood, TEA has a strong odor.

### TROUBLESHOOTING

Flow rates of > 0.5 mL/min in the load step (step 3) will cause sample breakthrough which reduces oligonucleotide recovery in final elution (step 7).

### RECOVERY CALCULATION

Recovery of target oligonucleotide is determined by analysis with a UV absorbance spectrometer.

Take 10 μL of sample solution (prior to loading), dilute to 1 mL and measure Absorbance  $A_{260}(L)$ .

Take 10 μL of final elution (step 7), dilute to 1 mL and measure absorbance  $A_{260}(E)$ .

$V_E$  = elution volume from step 7

$V_L$  = elution volume from step 3

$$\text{Recovery (\%)} = \left[ \frac{A_{260}(E)}{A_{260}(L)} \times 100 \right] \times \frac{V_E}{V_L}$$

DMT = dimethoxytrityl  
ACN = acetonitrile  
TEAA = triethylamine acetate  
TFA = trifluoro acetic acid

### OLIGONUCLEOTIDE PURITY DETERMINED BY CAPILLARY GEL ELECTROPHORESIS.

Source: M. Gilar, E.S.P. Bouvier, J. Chromatography A, vol 890 (1), 167-177.